The speed of sound c is defined by

$$c = \sqrt{\frac{\gamma RT}{M}} , \qquad (A1)$$

where  $\gamma$  is the adiabatic index, R is the universal molar gas constant (~ 8.3145 J/mol/K), T is absolute temperature (K) and M is the molar mass of gas.

From equation (A1), the speed of sound in dry air ( $\gamma$ =1.402, M=28.966×10<sup>-3</sup> kg/mol) is approximated as

$$c = 331.5 + 0.6t$$
, (A2)

where t is temperature in degrees Celsius ( $^{\circ}C$ ).

The speed of sound for gas-gas mixture is defined by

$$c_{mix} = \sqrt{\gamma_{mix} R_{mix} T} . \tag{A3}$$

The definitions of  $\gamma_{mix}$  and  $R_{mix}$  are as below when gas A and gas B are mixed.

$$\gamma_{mix} = x_A \gamma_A + x_B \gamma_B, \qquad (A4)$$
$$R_{mix} = x_A R_A + x_B R_B = x_A \frac{R}{M_A} + x_B \frac{R}{M_B}, \qquad (A5)$$

where x is the mass fraction,  $\gamma_A$  and  $\gamma_B$  are the adiabatic indexes for each gas and  $R_A$  and  $R_B$  are molar gas constants for each gas.

## [Reference]

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